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APPLICATION OF THE DEFINITE INTEGRAL TO CHEMICAL TECHNOLOGY

Annotation:	Precise. integrals and their chemistry to technology implementation is very wide and many edge topic be, chemistry and technological processes
	mathematician in terms of in modeling very important importance has.
	Determined integrals mainly materials and energy of the flow distribution, reactions kinetics, diffusion processes and other various chemical events
	analysis in doing is used. Below clear integrals chemistry to technology of
	application some main directions seeing will be released.
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Keywords:	integral, definite integral, chemical reaction, reaction speed, concentration, time, reactor, chemical reactor, diffusion.
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1. Chemical Reactions Model to do.

Chemistry in technology clear integrals, chemical reactions kinetics and mechanics according to model in creation Chemical reactions in learning, especially in reaction speed and catalytic processes important importance has.

A. Reaction speed and concentration.

Chemical reaction speed usually concentrations to time related change with is described. If any if the reaction is $A \rightarrow B$ and reaction speed as follows if given:

$$r = -k[A]^n$$

This on the ground:

- \triangleright r—reaction speed,
- \triangleright k— reaction rate coefficient,
- \triangleright [A]— Aconcentration of the substance,
- > n— reaction order.

If If A the concentration [A](t) of a substance depends on time, then by using the definite integral, we can find the time dependence of this concentration . change calculation For example, if the reaction order first level if, then integration as follows will be:

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$$\frac{d[A]}{dt} = -k[A].$$

This equation integration through concentration to time related change find possible:

$$[A](t) = [A]_0 e^{-kt}$$
.

This formula is chemical reaction during substances concentration to time related change represents.

B. Catalytic Reactions.

Catalysts work mechanism also clear in learning integrals Catalytic in processes, for example, reactions sequence or reaction and catalyst mutual impact modeling integral calculations for necessary.

2. Diffusion and Massani Distribution.

Diffusion processes chemistry in technology very important place holds, because many processes, for example, chemical reactors, mutual effects or absorption and adsorption processes diffusion through done is increased.

A. Diffusion equations.

Diffusion processes, for example, Fick's law is based on:

$$J = -D\frac{\partial C}{\partial x}.$$

This on the ground:

- > I— substance flow (mass distribution rate),
- > D— diffusion coefficient,
- > C—concentration,
- \rightarrow x—distance in space.

If the concentration time and into space related if so, this equations using definite integral is solved. For example, if diffusion process to time looking at change need if, Fick's second law as follows is written:

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2}.$$

This equation can be solved using integrals. solution through substances spread to time related change calculation possible.

B. Mass General Flow.

Chemistry in technology of the mass general flow, for example, reactors and related in systems, many processes by controlled. Mass. spread study for integral application necessary, because this in systems one or one how many components together diffusion and reactions there is.

3. Thermodynamics and Energy Flow.

Chemistry in technology energy currents calculation for clear of the integral application very important. Thermodynamics processes, especially enthalpy, enthalpy change, and work to perform such as energy in modeling changes integrals is used.

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A. Energy flow and heat exchange.

Chemistry in technology heat exchange processes, such as heaters, condensers, and in evaporators energy balance calculation for need a definite integral If the system heat flow time and to phase related if so, it clear integration through analysis to do possible.

B. Reaction heat.

Chemical reactions heat change calculation for reaction enthalpy integral to take need to be possible. For example, exothermic or endothermic reaction during of the system heat change using definite integral is found.

4. Reactor Design and Optimal Performance.

Chemistry in technology, especially reactors and chemical working release in the processes clear integral application, system efficiency maximum to the level lift for It is necessary. the following includes:

A. Reactors for materials and energy balance.

Reactor in systems, for example, continuous and party in reactors, material and energy balances Integral is used in calculation. Through this in the system of substances flow, reaction speed and thermodynamic changes determination possible.

B. Optimal operating conditions.

Reactor optimal work in design conditions (e.g., temperature, pressure, and concentration (determination) for clear integrals using of the system work activity optimization possible. These conditions the reactor maximum in efficiency use for will be necessary.

5. Ovens and Heat Exchange In the processes.

Furnaces, distillation columns and other heat exchange in systems of heat spread and energy flow in calculation using definite integral following issues solution possible:

- **Heat exchange Speed**: On devices energy exchange and heat Modeling change.
- ➤ **Maximum efficiency find**: Chemical working release processes heat flow with how connected and this streams effective to manage to determine.

Conclusion. Precise integrals chemistry in technology very important tool is different processes mathematician in terms of in modeling They are used. chemical reactions kinetics, mass distribution processes, energy balance and thermodynamic changes analysis in doing help Chemistry in technology clear integral use, processes optimization and efficiency increase for It is necessary.

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