

Synthesis and Characterization and Biological Study of Some Oxazepine Derivatives from 2-Acetylthiazole

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Abstract

This study involved the preparation of different α - β unsaturated compounds by reacting 2-acetylthiazole with different aromatic benzaldehyde substituents (E1-E5) and reacting compounds (E1-E5) with 2-aminophenol. The minimal medium with 10% NaOH was then examined for the inhibitory activity against E. coli and Gram-positive Staphylococcus aureus using some spectroscopic methods. : H1-NMR, 13C-NMR, and TLC techniques were used to confirm the effectiveness of the prepared compounds.

Keywords: Chalcone, Oxazepines, biological activity.

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1. Introduction

Chalcone is composed of two aromatic rings connected by an α - β -unsaturated carbonyl system, which is a derivative of 1,3-diphenyl-2-propen-1-one [1-3]. Nucleophilic addition reactions form a colored crystalline compound that is insoluble in water but soluble in organic solvents. The importance of these compounds lies in the successive presence of two functional groups, the double bond and the carbonyl group [4-7]. **Oxazepines** are heptameric compounds called oxazepines if unsaturated but oxazepines if saturated [8,9]. They consist of five carbon atoms and two heteroatoms [10]. Isomers, depending on the position of the oxygen and nitrogen atoms in the heptameric structure [11]. Many studies have shown that oxazepines can treat psychiatric disorders and many diseases [12] and have many medical and pharmaceutical applications. They also have anticancer activity, antioxidant activity, and activity against bacteria and microorganisms [13,14]

2. Materials and Methods:

2.1. Chemicals used: Utilized were chemicals produced by BDH Thomas, Fluka, Merck, and Aldrich.

2.2. Preparation of Chalcone [15]

(2-acetylthiazole) (0.007 mol, 1 g) was dissolved in ethanol (10 ml), a solution of (10% NaOH) (10 ml) was added, and stirred for 10 minutes, then (0.007 mol) aromatic aldehyde was added and dissolved in (10 ml of absolute ethanol) and the mixture was stirred in a water bath at a temperature of (20-40 °C) for (2-3) hours and the mixture was left in the refrigerator overnight and then added to crushed ice to form a precipitate that was filtered, collected and recrystallized from ethanol. As in Table 1

2.3. Preparation of Oxazepine .[16]

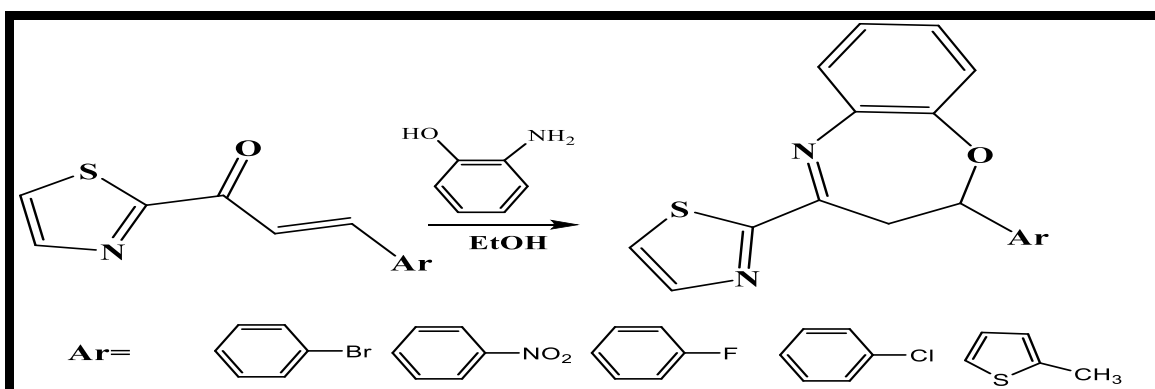
Dissolve an equal amount of one of the chalcones (0.00257 mol, ~0.691 g) in (10 mL) of absolute ethanol and add an excess of aminophenol dissolved in (5 mL) of ethanol and stir for (10) minutes, then add 5 mL of 10% sodium hydroxide solution, stir continuously, heat for (5-8) hours, then concentrate the solution, cool, add to crushed ice, add dilute hydrochloric acid dropwise to neutralize, use a TLC plate to follow the reaction process, as shown in Table 1.

.2.4. study of Biological activity

Dissolve Mueller Hinton agar (39 g) in (1 L) of distilled water, then heat and dissolve it using a magnetic stirrer, then sterilize it using an autoclave at a temperature of (121 °C). °C and a pressure of (1.5 bar) for 2 hours, then cool to 50 °C, then pour into Petri dishes and freeze at room temperature[17,18]. Two types of bacterial isolates were tested. 1 Gram-negative strain [Gr-ve] E. coli, 1 Gram-positive strain [Gr+ve] Gold Staphylococcus aureus[19,20]. Two colonies of pure bacterial isolates of both Gram-positive and Gram-negative bacteria were transferred from the solid culture medium to test tubes containing (5 ml) distilled water using heat-sterilized holders. The tubes were incubated at 30 °C. (37°C) for (16-20) hours, then diluted with physiological solution until the turbidity reaches a level similar to the standard turbidity to obtain a cell count of approximately (1.5 x 10⁸) cells/ml[21-25].

3. Results and discussions

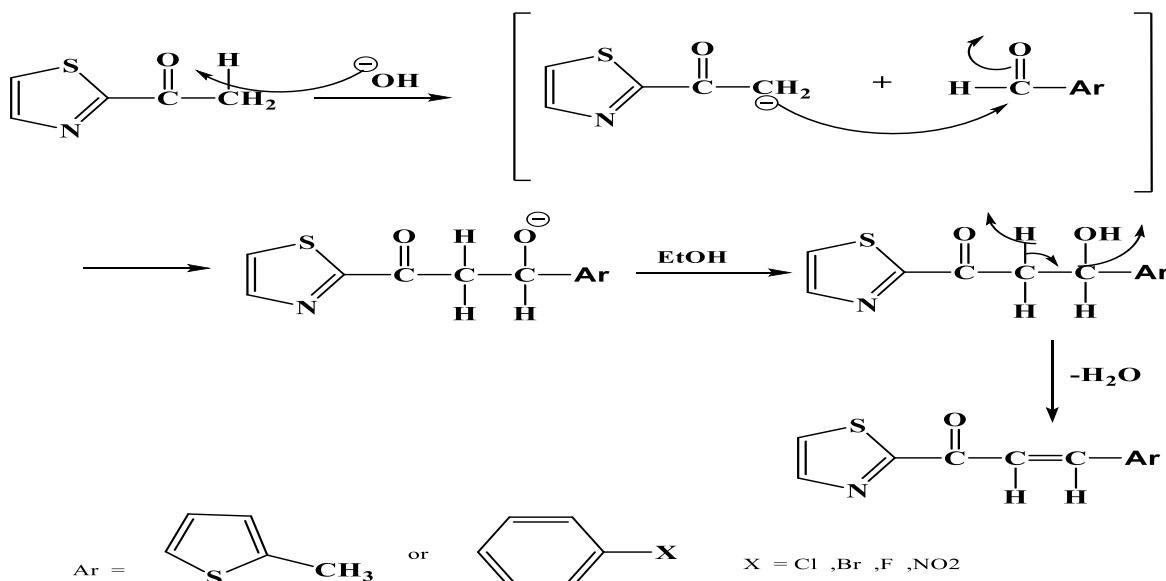
The following Scheme shows the preparation of azetidene derivatives.



Scheme 1: Shows the path of the prepared compounds.

3.1. Characterization of Chalcone.

Proposed mechanism for the following reaction:



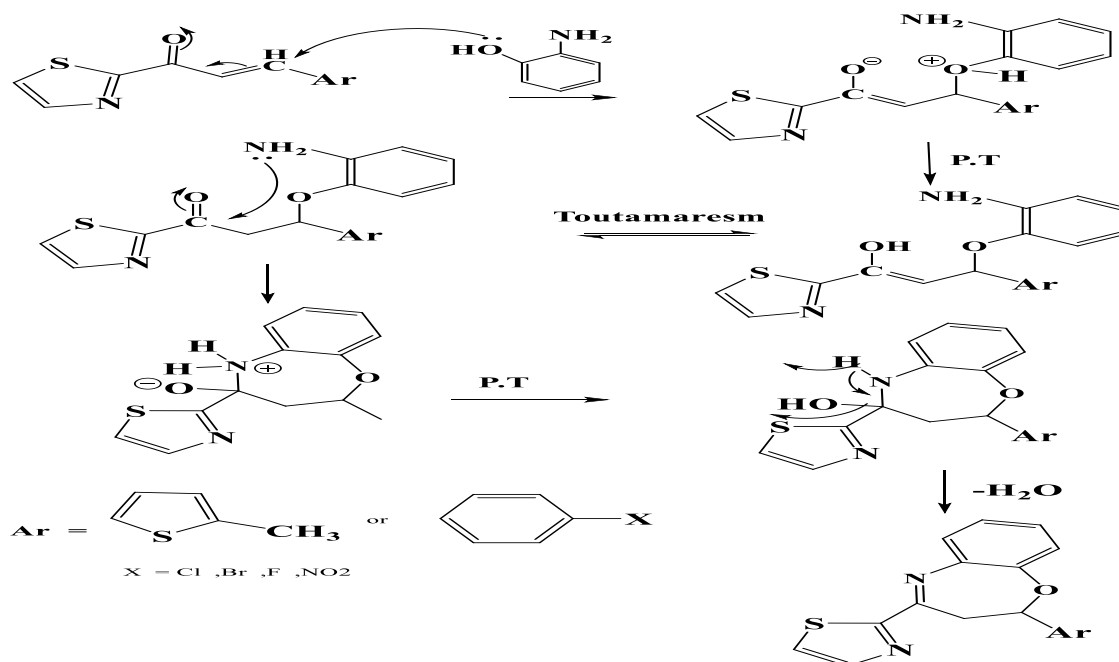
Scheme (2): shows the mechanism of formation of chalcone compounds (E1-E5)

When studying the ¹H-NMR spectrum of compound [E4], a double signal was observed at ppm (6.82-6.85) attributed to α (=CH) and a double signal at ppm (7.04-7.06) attributed to β (=CH). A multiple signal at ppm (7.26-7.86) was attributed to the aromatic rings[26]. As in Figure 1

The ¹³C-NMR spectrum of compound [E4] using the solvent (DMSO-d₆) showed a signal at the ppm position (187.55) attributed to the carbonyl carbon (C=O), at the ppm position (120.37) a signal attributed to the carbon of the (α-CH) group adjacent to the benzene ring at the point (144.45), a signal attributed to the carbon of the (β-CH) group of the alkene group appeared at the ppm position (144.45), and at the ppm position multiple signals attributed to the aromatic carbons appeared at (128.16-142.86), as shown in Figure 2

3.2. Characterization of Oxazepine.

Proposed mechanism for the following reaction:



Scheme (3): Shows the mechanism of formation of oxazepine rings (E11-E15)

The ¹H-NMR spectrum of the compound (E11) showed a double signal at (2.96, 2.99) ppm attributed to (CH₂) of the oxazepine ring, a triple signal at (3.74-3.79) ppm attributed to (-CH-) of the oxazepine ring, and the signals at the position (7.04-7.97) ppm belong to the aromatic system[27], as in Figure 3.

When studying the ¹³C-NMR spectrum of compound (E11), a signal was observed at ppm (21.61) due to carbon (CH₂) of the oxazepine ring, a signal at ppm (75.11) due to carbon (-CH- O) in the oxazepine ring, a signal at ppm (162.97) due to carbon (C=N) of the oxazepine ring, and signals at ppm (113.34-144.80) due to the aromatic system, as in Figure 4.

3.2. Evaluation of the Biological Activity of Prepared Compounds

Compounds with heterocyclic rings have different biological activities against Gram-positive and Gram-negative bacteria[28,29]. The bioavailability of some of the prepared compounds was evaluated in this study on two types of bacteria: Escherichia coli (negative) and Staphylococcus aureus. These bacteria were chosen for their medical importance, as they cause a range of diseases. They have varying degrees of resistance to antibiotics. The bioavailability of the prepared compounds was evaluated using the etching method (26) and the inhibition level (zone of inhibition) was measured. The results showed that the prepared compounds could inhibit the growth of antibiotics[30-33]. The bacteria bind to Gram-positive and Gram-negative staining types in different proportions. Using the antibiotic ampicillin as a control group, compounds (E1, E4, and E11) had significant inhibitory effects on E. coli infection, and compounds (E14, E12, and E5) had significant inhibitory effects on Staphylococcus aureus[34,35]. The relationship is directly related to the concentration used and the inhibitory effect of the compound. The highest degree of inhibition occurs at a concentration of 0.01 mg/ml[36-38]. as shown in the table2

Table (1): Some physical properties of for Prepared compounds (E1-E15).

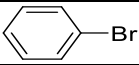
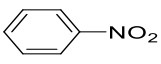
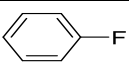
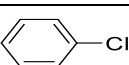
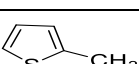
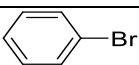
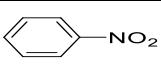
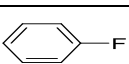
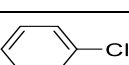
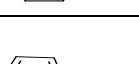
Comp. No.	Ar	Molecular Formula	m.p. °C	Yield%	Color
E1		C ₁₂ H ₈ BrNOS	107-110	94	off white
E2		C ₁₂ H ₈ N ₂ O ₃ S	119-121	79	Dark brown
E3		C ₁₂ H ₈ FNOS	114-116	75	Light Yello
E4		C ₁₂ H ₈ ClNOS	125-127	80	Light brown
E5		C ₁₀ H ₉ NOS ₂	94-96	68	Dark brown
Comp. No.	Ar	Molecular Formula	m.p. °C	Yield%	Color
E11		C ₁₈ H ₁₃ BrN ₂ OS	205-207	68	Brown
E12		C ₁₈ H ₁₃ N ₃ O ₃ S	270-272	71	Dark brown
E13		C ₁₈ H ₁₃ FN ₂ OS	159-161	62	Dark brown
E14		C ₁₈ H ₁₃ ClN ₂ OS	144-146	74	Liaghtbrown
E15		C ₁₇ H ₁₄ N ₂ OS ₂	154-156	55	Brown

Table (2): Biological efficacy of produced substances and control methods (measured in millimeters of inhibition).

Comp. No	Conc. mg/ml	<i>Staphylococcus aureus</i>	<i>Escherichia coil</i>
E1	0.01	31	15
	0.001	21	13
	0.0001	10	9
E4	0.01	26	26
	0.001	19	17
	0.0001	13	16
E5	0.01	10	Niz
	0.001	8	6
	0.0001	5	10
E11	0.01	15	28
	0.001	13	17
	0.0001	12	16
E12	0.01	Niz	10
	0.001		0
	0.0001		0
E14	0.01	Niz	13
	0.001		12
	0.0001		12
Ampicillin	For every focus.	16	15

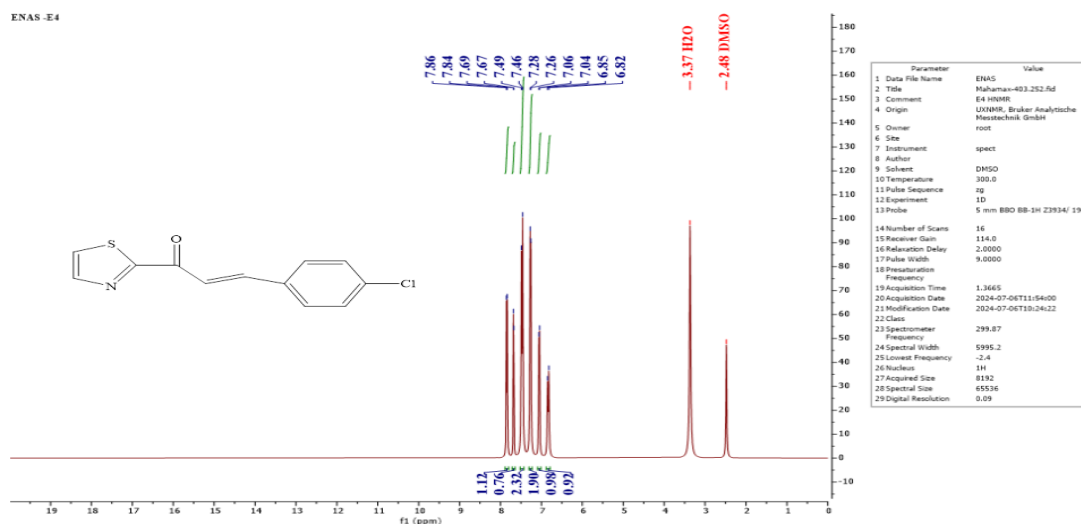


Figure (1): 1-H NMR spectra of the substance (E4).

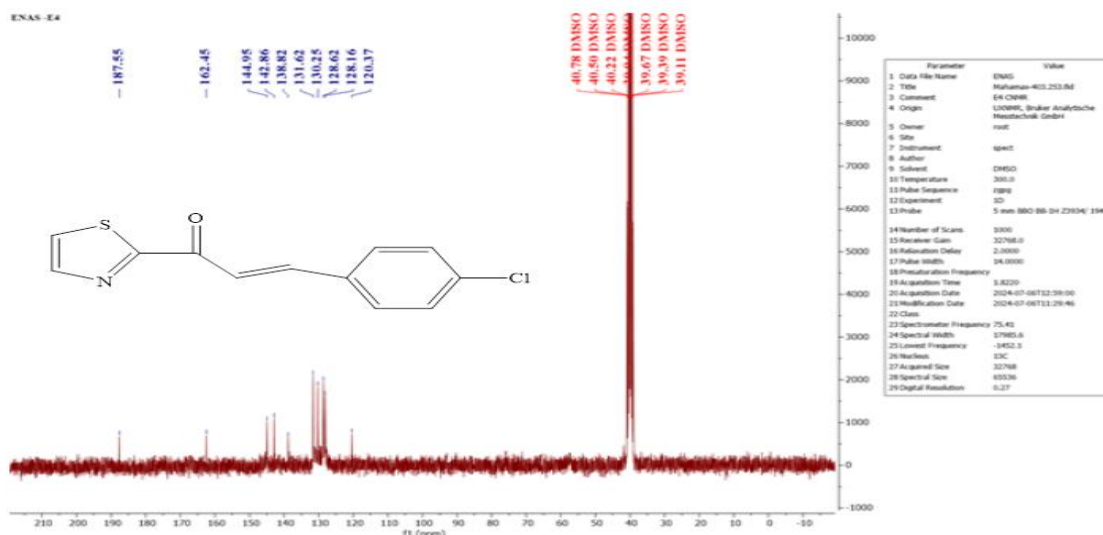


Figure (2): ¹³C- NMR spectra of the substance (E4).

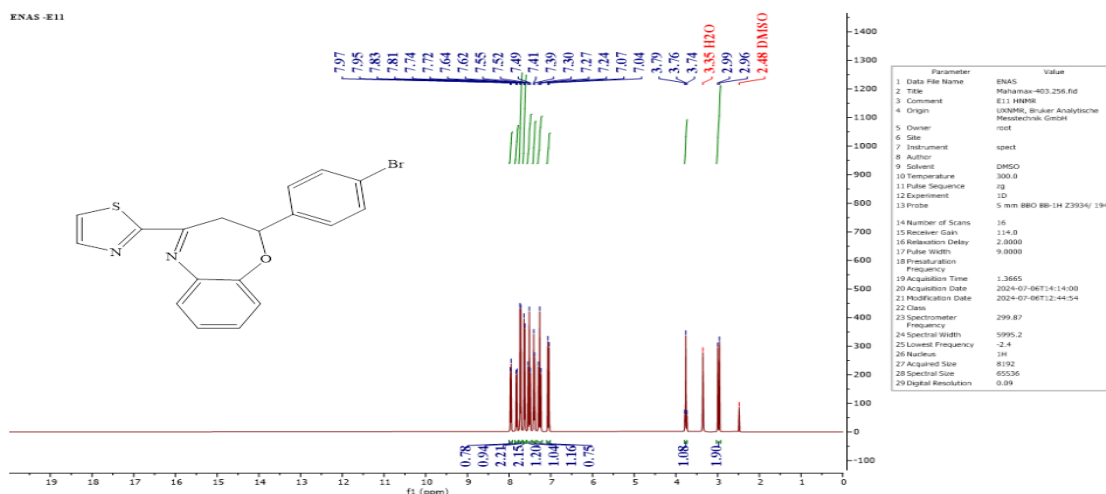


Figure (3): ¹H NMR spectra of the substance (E11).

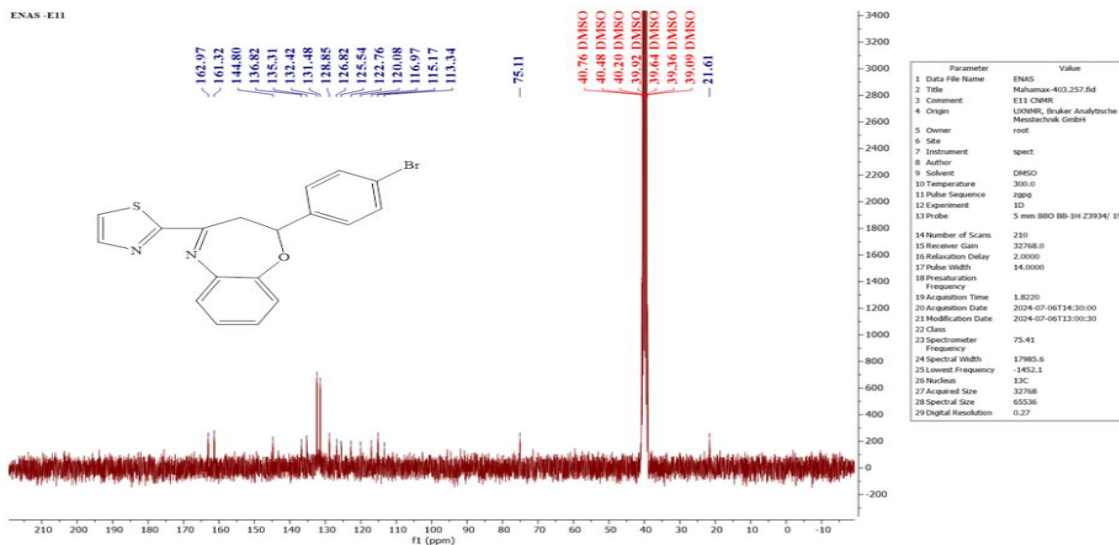


Figure (4): ¹³C- NMR spectra of the substance (E11).

4. Conclusions

The prepared compounds showed high purity as proven by spectroscopic measurements such as proton and carbon nuclear magnetic resonance spectra and showed varying effectiveness against the two types of bacteria used, as some of them showed effectiveness against the first type and others showed effectiveness against the second type.

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